







20 protons

$$1.6 \times 10^{-19} \text{ C}$$

$$Q = 32 \times 10^{-19} \text{ C}$$

$$100 \text{ nm}$$
$$100 \times 10^{-9}$$



$$r = 100 \times 10^{-9} \text{ m}$$



1 electron

$$E = \frac{kQ}{r^2}$$

$$2.88 \times 10^6 \frac{\text{N}}{\text{C}}, \text{ RIGHT}$$

e^- ~~~~~

$$4.6 \times 10^{-13} \text{ N}$$

LEFT