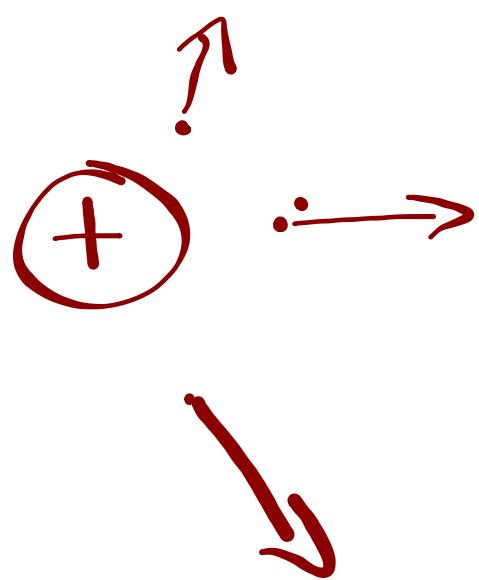
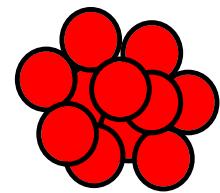


-42 μC
 3N/C, LEFT
 $F = E q$
 $0.07 \frac{\text{N}}{\text{C}}, \text{RIGHT}$
18 μC





20 protons

100 nm
 100×10^{-9}

$$r = 100 \times 10^{-9} \text{ m}$$

$$1.6 \times 10^{-19} \text{ C}$$

$$Q = 32 \times 10^{-19} \text{ C}$$



1 electron

$$\underline{E} = \frac{kQ}{r^2}$$

$2.88 \times 10^6 \frac{N}{C}$, RIGHT

$e^- 4.6 \times 10^{-13} N$

LEFT